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Original Article

Green synthesis and characterization of silver nanoparticles from *Ducrosia flabellifolia* Boiss. aqueous extract: Anti-quorum sensing screening and antimicrobial activities against ESKAPE pathogens





¹Department of Biology, College of Science, Hail University, P.O. Box 2440, Ha'il 2440, Saudi Arabia

²Laboratory of Genetics, Biodiversity and Valorization of Bio-Resources (LR11ES41), Higher Institute of Biotechnology of Monastir, University of Monastir, Avenue Tahar Haddad, BP74, Monastir 5000, Tunisia

³Medical and Diagnostic Research Center, University of Ha'il, Hail 55473, Saudi Arabia

⁴Chemistry Department, Faculty of Science, Ha'il University, 81451 Hail, P.O. Box 2440, KSA

⁵Laboratory of Water, Membranes and Environmental Biotechnologies, Center of Research and Water Technologies, P. B 273, Soliman 8020, Tunisia

⁶Department of Clinical Nutrition, College of Applied Medical Sciences, University of Hail, Hail PO Box 2440, Saudi Arabia ⁷Research and Development Cell, Department of Biotechnology, Parul Institute of Applied Sciences, Parul University, Vadodara, 391760, India ⁸Department of Chemistry, College of Science and Humanities in Al-Kharj, Prince Sattam bin Abdulaziz University, Al-Kharj, Saudi Arabia ⁹Faculty of Science and Arts in Baljurashi, Al-Baha University, P.O. Box 1988, Al-Baha 65527, Saudi Arabia

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Abstract

Biosynthesis of silver nanoparticles using natural compounds derived from plant kingdom is currently used as safe and low-cost technique for nanoparticles synthesis with important abilities to inhibit multidrug resistant microorganisms (MDR). ESKAPE pathogens, especially MDR ones, are widely spread in hospital and intensive care units. In the present study, AgNPs using Ducrosia flabellifolia aqueous extract were synthesized using a reduction method. The green synthesized D. flabellifolia-AgNPs were characterized by UV-Vis spectrophotometer, Scanning electron microscopy (SEM), and X-ray diffraction assays. The tested D. flabellifolia aqueous extract was tested for its chemical composition using Liquid Chromatography-Electrospray Ionization-Mass Spectrometry (LC-ESI-MS). Anti-quorum sensing and anti-ESKAPE potential of D. flabellifolia-AgNPs was also performed. Results from LC-ESI-MS technique revealed the identification of chlorogenic acid, protocatechuic acid, ferulic acid, caffeic acid, 2,5-dihydroxybenzoic acid, and gallic acid as main phytoconstituents. Indeed, the characterization of newly synthetized D. flabellifolia-AgNPs revealed spherical shape with mean particle size about 16.961±2.914 nm. Bio-reduction of silver was confirmed by the maximum surface plasmon resonance of D. flabellifolia-AgNPs at 430 nm. Newly synthetized D. flabellifolia-AgNPs were found to possess important anti-ESKAPE activity with low minimal inhibitory concentrations (MICs) ranging from 0.078 to 0.312 mg/mL, and low minimal bactericidal concentrations (MBCs) varying from 0.312 to 0.625 mg/mL. Moreover, D. flabellifolia-AgNPs were active against Candida utilis ATCC 9255, C. tropicalis ATCC 1362, and C. albicans ATCC 20402 with high mean diameter of growth inhibition at 5 mg/ mL, low MICs, and minimal fungicidal concentrations values (MFCs). The newly synthetized D. flabellifolia-AgNPs were able to inhibit violacein production in Chromobacterium violaceum, pyocyanin in Pseudomonas aeruginosa starter strains. Hence, the newly synthesized silver nanoparticles using D. flabellifolia aqueous extract can be used as an effective alternative to combat ESKAPE microorganisms. These silver nanoparticles can attenuate virulence of Gram-negative bacteria by interfering with the quorum sensing system and inhibiting cell-to-cell communication.

Keywords: *Ducrosia flabellifolia* Boiss.; Silver nanoparticles; Chemical composition; Anti-ESKAPE; Quorum sensing.

1. Introduction

Infectious diseases continue to be an increasing factor of illness and mortality worldwide, reinforced by high resistance of bacterial, mainly bacterial, and viral infections to drugs [1-3]. In addition, the misuse of antibiotics facilitates the development of multidrug-resistant organisms, as well as the appearance of new pathogens against which there is still no treatment [4, 5]. Nowadays, nanomedicine is a rapidly developing field that contributes to the production of a wide range of various synthesized metal

^{*} Corresponding author.

E-mail address: m.snoussi@uoh.edu.sa (M. Snoussi).

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nanoparticles (MNPs) that explore the possibilities of their uses for the prevention, treatment, diagnosis, and control of disease [6-8]. Additionally, MNPs provide solutions to environmental challenges and are able to surmount some essential problems of conventional small molecules or biomacromolecules (e.g., DNA, RNA, and proteins) and play a dynamic role in introducing various approaches for combating disease, increasing disease diagnostics, and developing new measures for manipulation of plants and pathogens [9].

For instance, application of MNPs has gained tremendous attention due to their specific physiochemical properties such as size, distribution, and morphology, increased electrical conductivity, roughness and the ability to strengthen metals and alloys [10-12]. They have been also studied for their catalytic activity, magnetic, optical, and electronic properties, and antibacterial activity, as well as they are used as assembling of innovative functional materials in medicine, engineering, environment, and agriculture [13-15]. Amongst them, silver nanoparticles (AgNPs) with a diameter of less than 100 nm have gained boundless interest and remain the most noble metal in fabrication of nanoparticles, especially in the field of health and medicine [16]. They are extremely important due to their attractive properties and wide spectrum of anti-inflammatory effects, curing wound healing, microbial proliferation, bactericidal and fungicidal activities as well as their ability to be used in biomedical device coatings, drug-delivery carriers, imaging probes, and diagnostic and optoelectronic platforms [17, 18]. It has been reported that silver nanoparticles possess stronger antibacterial capabilities and are used as anticancer agents [19, 20]. Siver is routinely used in the form of silver nitrate (NO,⁻) for antimicrobial activity. In fact, several multidrug-resistant microorganisms (MDR) circulating in clinical and hospital environments are responsible for several main infections and human diseases (Van Melderen and DeBast, 2009). The emergence of these MDR pathogens can be explicated by the overuse of antibiotics in healthcare, animal, and agriculture sectors [21, 22]. These MDR bacteria are known as ESKAPE pathogens and include Enterococcus faecium, Staphylococcus aureus, Klebsiella pneumoniae, Acinetobacter baumannii, Pseudomonas aeruginosa, and Enterobacter spp. [23-25]. These microorganisms are known to produce exopolysaccharides and are able to adhere to several biotic and abiotic surfaces [26].

Plant-mediated synthesis of nanomaterials known for its eco-friendly nature and cost-effectiveness, aims at decreasing the usage of toxic chemicals, providing a better platform for nanoparticles synthesis since they are safe and contain reducing and capping agents [27, 28]. Extract Ag nanoparticles are important to investigate mainly because they can be used in nanomedicine and are easily packaged in the form of infusions/fluids [16]. This has fascinated scientists to study the way metal ions uptake and bio-reduction by plants. Among various plants, we have chosen Ducrosia flabellifolia from Hail region aerial parts extracts for the present study since they have been proven for their potent pharmacological effect [29]. Hence, we newly synthesized silver nanoparticles using D. flabellifolia aqueous extract. The chemical composition of the obtained aqueous extract was performed using LC-ESI-MS/MS technique. The characterized silver nanoparticles were tested for their abilities to inhibit the growth of ES-

KAPE pathogens and Candida species. Moreover, the antiquorum sensing activities were tested using three starter strains (*Chromobacterium violaceum*, *Pseudomonas aeruginosa*, and *Serratia marcescens*).

2. Material and Methods

2.1. Plant material sampling

Aerial parts from *D. flabellifolia* [30, 31] were collected from the Hail region (Saudi Arabia) in October 2019. For the experiment, 100 g of fresh aerial parts from *D. flabellifolia* plant species (Fig. 1) were grounded and mixed in distilled water (10g plant material+ 100 ml of distilled water), then filtered using Whatman N°1. The filtrate was kept in a dark glass bottle at -4°C till use.

2.2. Phytochemical analysis

The previously validated method was used for the analysis of phenolic compounds by LC-ESI-MS/MS [32]. Multiple reaction monitoring transitions, the optimum collision energies, and retention times for each species, representative LC-ESI-MS/MS chromatograms of phenolic compounds, and calibration curves, and sensitivity properties of the method are previously detailed by Snoussi and colleagues in 2022 [30].

2.3. Synthesis of nanoparticles

Silver Nitrate (AgNO₃) was used for the reaction [33]. For the experiment, (*D. flabellifolia*)-AgNPs were prepared by the green method using the aqueous extract of *D. flabellifolia* (AEDF) at 80°C for 15 mn. In a typical reaction mixture, 100 mL of a 3 mM Silver Nitrate (AgNO₃) solution was put in a two-neck round bottom flask containing 197 mL of bi-distilled (BD) water. The resulting mixture was heated under reflux at 80°C and constant stirring. Once the set temperature was reached, 3 mL of the aqueous extract of *D. flabellifolia* were added. When the set temperature was reached again, the count of the duration of the reaction was started. Under the above conditions, the concentration of AgNO₃ in the reaction mixture was 1 mM and each 100 mL of the reaction mixture contains 1 mL of the AEDF.

2.4. Characterization of the obtained AgNPs 2.4.1. *Ultraviolet-Visible Spectroscopy*

The formation of AgNPs was confirmed by UV-Vis spectroscopy (UV-Vis). The Surface Plasmon Resonance



Fig. 1. Ducrosia flabellifolia plant body [31].

(SPR) is the characteristic property of silver, gold and some other metal nanoparticles. UV-Vis spectra were made using a Spectro UV-Vis Double Beam PC Scanning Spectrophotometer (Labomed UVD-2950, USA).

2.4.2. Scanning Electron Microscopy (SEM)

The morphology of the prepared (*D. flabellifolia*)-AgNPs was examined by using Field Emission Scanning Electron Microscope (SEM) (Thermo Scientific Quattro, Waltham, MA, USA). SEM imaging was performed under a high vacuum and samples (fragments of the prepared glass slide) were metalized with gold prior to characterization. SEM images were analyzed by using Image J software to measure the particle size and size distribution of the prepared AgNPs.

2.4.3. X-Ray Diffraction (XRD)

The identification of the crystalline structure of the prepared (*D. flabellifolia*)-AgNPs was made by using an X-Ray Diffractometer (Shimadzu, XRD-7000 with an X-ray wavelength Cu detector, Japan). The scanned 2 Theta ranges between 5 and 80 degrees.

2.5. Biological activities of the synthesized AgNPs

2.5.1. *Antimicrobial activities of (D. flabellifolia)-AgNPs* Well diffusion assay was used for the determination of the diameter of growth inhibition zone estimated on agar medium (Mueller Hinton for bacteria and Sabouraud Chloramphenicol agar for *Candida* strains). This study includes 13 ESKAPE pathogens and 3 Candida species as listed in Table 1 below.

After incubation, the diameter of the growth inhibition zone (GIZ) was recorded for each bacterial and yeast strain and the mean diameter was calculated as the average of three records. All tests were done in triplicate. Ampicillin (10 UI) and amphotericin B (10 mg/mL) were used as control drugs.

Minimal inhibitory concentration (MICs) and minimal bactericidal/fungicidal concentration (MBC/MFC) were determined in liquid broth using microdilution assay. *D. flabellifolia*-AgNPs were tested at different concentrations ranging from 0.024 mg/mL to 2.5 mg/mL. The character of the tested solution was estimated by using the scheme proposed by Moroh and colleagues [34].

2.6. Anti-quorum sensing activities of *D. flabellifolia*-AgNPs

2.6.1. Quorum sensing inhibition assay

A well-diffusion assay was used to screen the anti-QS activity of different synthesized nanoparticles against *Chromobacterium violaceum* MTCC-2656, *Pseudomonas aeruginosa* MTCC-741, and *Serratia marcescens* MTCC-97 [35]. Briefly, overnight-grown culture of bacterial strain was streaked onto Lauria Bertani agar plates and wells were made with gel puncture. Following puncture of the wells, 50 μ L of different synthesized nanoparticles were inoculated into the respective wells and the plates were incubated at 30°C for 24 h. An inhibitory effect of the indicator strain surrounding the well was considered a positive sign of QS interference.

2.6.2. Determination of Minimum Inhibitory Concentration (MIC)

In order to determine the MIC values of newly synthe-

sized nanoparticles against different pathogenic bacteria such as *C. violaceum* MTCC-2656 (*C. violaceum*), *P. aeruginosa* MTCC-741, and *S. marcescens* MTCC-97, broth dilution method was used [36]. A series of double dilutions of different synthesized nanoparticles from 1000 μ g/mL to 1.95 μ g/mL concentrations were used in Mueller Hinton broth with an active bacterial culture. A control was prepared using only inoculated broth, which was incubated at 37 °C for 24 h. An MIC is defined as the lowest concentration that does not allow any visible growth on the tubes after the experiment.

2.7. Statistical analysis

All measurements will be carried out in triplicate and the results were presented as mean values \pm SD (standard deviations). Statistical analyses will be performed using a one-way analysis of variance ANOVA test.

3. Results

Table 2 summarizes the identified compounds in *D. flabellifolia* boiled extract using ESI-MS/MS technique. Results showed the identification of twenty-three compounds dominated (mg/g of extract) by chlorogenic acid (5980.96 \pm 73.12), protocatechuic acid (5980.96 \pm 73.12), ferulic acid (112.43 \pm 1.541), caffeic acid (63.61 \pm 0.495), 2,5-dihydroxybenzoic acid (60.68 \pm 2.518), and gallic acid (43.96 \pm 0.251).

The reduction of Ag+ ions in the aqueous solution was made by the reducing phytochemicals present in the boiled aqueous extract from *D. flabellifolia* aerial parts. The reduction was visible from the color change of the solution (Fig. 2). The reaction mixture was colorless or pale yellow for the most concentrated solutions at the beginning of the reaction and turned gradually to brownish once the reduction started.

The kinetic of *D. flabellifolia*-AgNPs formation was followed by UV-Vis. It can be seen from Fig. 3 that the wavelength of maximum absorption (λ max) corresponds to the Surface Plasmon Resonance (SPR) of the prepared *D. flabellifolia*-AgNPs. The SPR was small for this

Table 1. ESKAPE pathogens and Candida species used in this study.

Code	ESKAPE pathogens			
M2	Escherichia coli (212)			
M6	Escherichia coli (215)			
M7	Enterococcus faecalis (268)			
M8	Enterobacter cloacae (235)			
M9	Enterobacter faecium (260)			
M10	Acinetobacter baumannii (248)			
M11	Staphylococcus hominis (140 BC)			
M12	Staphylococcus aureus (259)			
M1	Staphylococcus aureus (217)			
M13	Staphylococcus epidermidis (BC 161)			
M14	Klebsiella pneumoniae (147)			
M15	Klebsiella pneumoniae (280)			
M16	Pseudomonas aeruginosa (249)			
Code	Candida spp.			
A1	Candida utilis ATCC 9255			
A8	Candida tropicalis ATCC 1362			
A15	Candida albicans ATCC 20402			

Plant silver NPs and their activities.

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N°	Identified compounds	Retention TimeAbondance(min)(mg/Kg of extract		Chemical Formula	Molecular Weight (g/mol)
1	Gallic acid	8.891	43.96±0.251	C ₇ H ₆ O ₅	170.12
2	Protocatechuic acid	10.818	295.35 ± 0.395	$C_7 H_6 O_4$	154.12
3	3,4-Dihydroxyphenylacetic acid	11.224	1.38 ± 0030	$C_8H_8O_4$	168.15
4	Pyrocatechol	11.506	9.66 ± 0.054	$C_6H_6O_2$	110.11
5	Chlorogenic acid	11.802*	4154.16±16.353	$C_{16}H_{18}O_{9}$	354.31
6	2,5-Dihydroxybenzoic acid	12.412	60.68±2.518	$C_7 H_6 O_4$	154.12
7	4-Hydroxybenzoic acid	12.439	20.82±1.182	$C_7 H_6 O_3$	138.12
8	(-)-Epicatechin	12.458*	3.88±0.154	$C_{15}H_{14}O_{6}$	290.27
9	Caffeic acid	12.841	63.61±0.495	$C_9H_8O_4$	180.16
10	Syringic acid	12.963	25.19±0.196	$C_{9}H_{10}O_{5}$	198.17
11	3-Hydroxybenzoic acid	13.259	7.76 ± 0.457	$C_7H_6O_3$	138.12
12	Vanillin	13.379	9.92±0.353	$C_8H_8O_3$	152.15
13	Sinapic acid	13.992	30.24±1.222	$C_{11}H_{12}O_{5}$	224.21
14	<i>p</i> -Coumaric acid	14.022	$30.772{\pm}1.458$	C ₉ H ₈ O ₃	164.16
15	Ferulic acid	14.120	112.43 ± 1.541	$C_{10}H_{10}O_{4}$	194.18
16	Luteolin 7-glucoside	14.266	3.87 ± 0.392	$C_{21}H_{20}O_{11}$	448.4
17	Hyperoside	14.506*	4.59±0.101	$C_{21}H_{20}O_{12}$	464.4
18	Rosmarinic acid	14.600	4.34±0.135	$C_{18}H_{16}O_{8}$	360.3
19	2-Hydroxycinnamic acid	15.031	22.00 ± 0.402	C ₉ H ₈ O ₃	164.16
20	Pinoresinol	15.118	4.42 ± 0.092	$C_{20}H_{22}O_{6}$	358.4
21	Eriodictyol	15.247	1.65 ± 0.103	$C_{15}H_{12}O_{6}$	288.25
22	Quercetin	15.668	1.99 ± 0.335	$C_{15}H_{10}O_{7}$	302.23
23	Kaempferol	16.236	1.48 ± 0.067	$C_{15}H_{10}O_{6}$	286.23

*: Compounds identified by positive Ionization mode

concentration at the beginning of the reaction. The λ max was around 430 nm-440 nm for the first twenty minutes.

Image J software was used to analyze SEM micrographs, in order to measure the particle size and thus the size distribution. In fact, more than one hundred nanoparticles were measured, and the size distribution was calculated automatically by the software. Results from SEM image analysis revealed that the shape of the newly synthetized *D. flabellifolia*-AgNPs was almost spherical with an average particle size of 16.961±2.914 nm (Fig. 4).

The XRD patterns are shown in Figure 5. The diffraction peaks were as follows, $2\theta = 27.75^{\circ}$, 32.16° , 38.05° , 44.14° , 46.15° , 54.72° , 57.49° , 64.40° , 67.27° , 74.20° , 77.26° and 77.48° . The following peaks, 38.05° , 44.14° , 64.40° , and 77.26° were assigned to the planes (111), (200), (220) and (311) corresponding to a cubic lattice of silver (Reference code: 00-004-0783, ICSD 64706 (PDF 87-597)). The following peaks 27.75° , 32.16° , 46.15° , 54.72° , 57.49° , 67.27° , 74.20° , and 77.48° were assigned to the planes (111), (200), (220), (311), (222), (400), (331), and (420) corresponding to a cubic lattice of silver chloride (Reference code: 01-085-1355, ICSD 064734 (PDF 31-1238)). The prepared nanoparticles are composed mostly of AgNPs (86%) and AgCl NPs (14%).

The newly synthesized *D. flabellifolia*-AgNPs were tested for their ability to inhibit the growth of twelve ES-KAPE pathogens and four *Candida* species. Results are summarized in Table 3 below. Overall, the newly synthesized nanoparticles were able to inhibit the growth of all tested ESKAPE pathogens and *Candida* species with different degrees and in a concentration-dependent manner.



Fig. 2. Color change of the reaction mixture maintained at 80°C after 70 min. (A): *D. flabellifolia* filtered. extract, (B): AgNO₃ solution, (C): mixture of *D. flabellifolia* and silver nitrate solution, and (D): reaction mixture after 70 mn.



Fig. 3. UV-Vis absorption spectra of the *D. flabellifolia*-AgNPs and the reaction mixture (DFAE $1\% (v/v) / 1 \text{ mM AgNO}_3$) after different reaction times at 80°C.





In fact, the highest mean growth inhibition zones (mGIZ) were obtained at 5 mg/mL of *D. flabellifolia*-AgNPs ranging from 15.33 ± 0.57 mm (For *Staphylococcus aureus*, 217) to 33.67 ± 1.15 mm (For *P. aeruginosa*, 249). The obtained mGIZ of *D. flabellifolia*-AgNPs at 1mg/mL, 2.5 mg/mL, and 5 mg/mL were higher than those of the standard antibiotic used (Ampicillin) for all tested bacteria, highlighting high antimicrobial activities of the newly synthetized *D. flabellifolia*-AgNPs.

Using microdilution assay, results have shown that *D. flabellifolia*-AgNPs were able to inhibit the growth of all tested ESKAPE pathogens at low concentrations ranging from 0.078 mg/mL (For *P. aeruginosa*) to 0.312 mg/mL (For *K. pneumoniae*, *S. epidermidis*, *E. cloacae*, and *E. coli*). Concentrations as low as 0.312 mg/mL of the newly synthetized *D. flabellifolia*-AgNPs are sufficient to kill all tested ESKAPE pathogens. The *D. flabellifolia*-AgNPs exhibited bactericidal profile against all tested ESKAPE pathogens with MBC/MIC ratio lower than 4. Moreover, MICs and MBCs values of *D. flabellifolia*-AgNPs were lower than the reference drug used (Ampicillin).

In addition, *D. flabellifolia*-AgNPs are less active against the tested *Candida* species with MICs of about 0.312 mg/mL, and MFCs values of about 5 mg/mL. The newly synthetized *D. flabellifolia*-AgNPs exhibited fungistatic profile, while the reference drug used has fungicidal action against all Candida species tested (Table 4).

3.1. Anti-QS activity

As part of the initial evaluation of the anti-QS activity of *D. flabellifolia*-AgNPs, their ability to modulate the QS activity of *C. violaceum*, *P. aeruginosa* and *S. marcescens* was examined *in-vitro*. An inhibitory effect of the indicator strain surrounding the well was considered a positive sign of QS interference. Results are presented in Fig. 6. To find out the antibacterial activity of *D. flabellifolia*-AgNPs against different pathogenic bacteria, a broth microdilution method was performed to determine the value of MIC. Results have shown that *D. flabellifolia*-AgNPs are able to inhibit *C. violaceum*, *P. aeruginosa*, and *S. marces-*



Fig. 5. PXRD diffractogram of the newly synthesized *D. flabellifolia*-AgNPs.

Table 3. Anti-ESKAPE and anti-Candida spp. activities of D. flabellifolia-AgNPs tested at three concentrations using well diffusion assay.

	Bacterial Strain	<i>D</i>	Amnicillin		
Code		mGIZ at mGIZ at		mGIZ at	mCIZ +SD
		1 mg/mL	2.5 mg/mL	5 mg/ml	
M2	Escherichia coli (212)	13.33 ± 1.15	15.67 ± 0.57	19.00 ± 1.00	6.00 ± 0.00
M6	Escherichia coli (215)	12.67±0.57	14.33 ± 0.57	$17.00{\pm}1.00$	$6.00 {\pm} 0.00$
M7	Enterococcus faecalis (268)	15.00 ± 1.00	16.67 ± 0.57	$29.00{\pm}1.73$	$6.00 {\pm} 0.00$
M8	Enterobacter cloacae (235)	11.33±1.15	15.67 ± 0.57	20.67 ± 0.57	$6.00 {\pm} 0.00$
M9	Enterobacter faecium (260)	12.33±0.57	17.33 ± 0.57	27.67 ± 0.57	$6.00{\pm}0.00$
M10	Acinetobacter baumannii (248)	14.33 ± 0.57	16.33 ± 1.15	22.33 ± 0.57	10.33 ± 0.57
M11	Staphylococcus hominis (140 BC)	12.67 ± 0.57	16.33 ± 0.57	22.67 ± 0.57	10.33 ± 0.57
M12	Staphylococcus aureus (259)	16.00 ± 0.00	19.67 ± 0.57	25.00 ± 0.00	$6.00{\pm}0.00$
$\mathbf{M}1$	Staphylococcus aureus (217)	12.33±0.57	13.67 ± 0.57	15.33 ± 0.57	14.00 ± 0.00
M13	Staphylococcus epidermidis (BC 161)	21.33±1.15	$25.00{\pm}1.00$	28.67±1.15	24.00 ± 1.00
M14	Klebsiella pneumoniae (147)	$15.00{\pm}1.00$	20.67 ± 0.57	23.67 ± 0.57	$6.00 {\pm} 0.00$
M15	Klebsiella pneumoniae (280)	12.67 ± 0.57	14.67 ± 1.15	18.67 ± 1.15	$6.00{\pm}0.00$
M16	Pseudomonas aeruginosa (249)	26.67±1.55	$30.33 {\pm} 0.57$	33.67±1.15	$6.00{\pm}0.00$
Code	Yeasts and molds	mGIZ at	mGIZ at	mGIZ at	Amphotericin B
Coue		1 mg/mL	2.5 mg/mL	5 mg/ml	GIZ ±SD
A1	Candida utilis ATCC 9255	6.00 ± 0.00	6.00 ± 0.00	12.67 ± 0.57	11.67 ± 0.57
A8	Candida tropicalis ATCC 1362	6.00 ± 0.00	$6.00 {\pm} 0.00$	12.67 ± 0.57	14.33 ± 0.57
A15	Candida albicans ATCC 20402	6.00 ± 0.00	$6.00 {\pm} 0.00$	13.33±1.15	12.67 ± 0.57

Table 4. Determination of MICs, MBCs, and MFCs values of *D. flabellifolia*-AgNPs tested against twelve ESKAPE and four Candida spp. strains using microdilution assay.

Cada	ESKAPE pathogens	D. flabellifolia-AgNPs			Ampicillin		
Code		MIC	MBC	MBC/MIC ratio	MIC	MBC	MBC/MIC ratio
M2	Escherichia coli (212)	0.156	0.625	4; Bactericidal	1.25	5	4; Bactericidal
M6	Escherichia coli (215)	0.3125	0.625	2; Bactericidal	1.25	5	4; Bactericidal
M7	Enterococcus faecalis (268)	0.156	0.325	2; Bactericidal	0.312	2.5	8; Bacteriostatic
M8	Enterobacter cloacae (235)	0.3125	0.625	2; Bactericidal	0.625	1.25	2; Bactericidal
M9	Enterobacter faecium (260)	0.156	0.312	2; Bactericidal	0.625	5	8; Bacteriostatic
M10	Acinetobacter baumannii (248)	0.156	0.625	4; Bactericidal	1.25	5	4; Bactericidal
M11	Staphylococcus hominis (140 BC)	0.156	0.625	4; Bactericidal	0.625	1.25	2; Bactericidal
M12	Staphylococcus aureus (259)	0.156	0.625	4; Bactericidal	0.625	1.25	2; Bactericidal
M1	Staphylococcus aureus (217)	0.156	0.625	4; Bactericidal	0.625	5	8; Bacteriostatic
M13	Staphylococcus epidermidis (BC 161)	0.3125	0.625	2; Bactericidal	0.312	0.625	2; Bactericidal
M14	Klebsiella pneumoniae (147)	0.3125	0.625	2; Bactericidal	0.625	5	8; Bacteriostatic
M15	Klebsiella pneumoniae (280)	0.156	0.625	4; Bactericidal	0.625	5	8; Bacteriostatic
M16	Pseudomonas aeruginosa (249)	0.078	0.625	8; Bactericidal	2.5	5	2; Bactericidal
Cada	Candida spp.	D. flabellifolia-AgNPs			Amphotericin B		
Code		MIC	MFC	MFC/MIC ratio	MIC	MFC	MFC/MIC ratio
A1	Candida utilis ATCC 9255	0.3125	5	16; Fungistatic	0.78	1.56	2; Fungicidal
A8	Candida tropicalis ATCC 1362	0.3125	5	16; Fungistatic	0.195	0.78	4; Fungicidal
A15	Candida albicans ATCC 20402	0.3125	5	16; Fungistatic	0.195	0.39	2; Fungicidal

cens at 125 µg/mL.

4. Discussion

In this study, we report for the first time the synthesis, characterization, antimicrobial, and anti-quorum sensing activities of silver nitrate nanoparticles by using boiled extract from D. flabellifolia aerial parts. The study of the phytochemical profile of the boiled aqueous extract revealed the identification of 23 molecules, mainly chlorogenic acid, protocatechuic acid, ferulic acid, caffeic acid, 2,5-dihydroxybenzoic acid, and gallic acid. Similar results have discussed the chemical composition of D. flabellifolia from Egypt, Jordan, Iran, and Saudi Arabia [30, 31; 37-41]. In fact, we have reported the identification of twenty phytocompounds in the methanol/water extract from D. flabellifolia aerial parts collected from Hail region (Saudi Arabia) where chlorogenic acid (5980.96 \pm 73.12 mg/g of extract), ferulic acid (180.58 \pm 2.77 mg/g of extract), caffeic acid (70.90 \pm 1.75 mg/g of extract), and sinapic acid (61.74 \pm 2.79 mg/g of extract) were the main compounds. In addition, our team has reported the identification of decanal and dodecanal as main compounds in D. flabellifolia essential oil collected from arid sandy soil in Hail (Saudi Arabia), and farnesyl pyrophosphate, methyl 7-desoxypurpurogallin-7-carboxylate trimethyl ether, dihydro-obliquin, gummiferol, 2-phenylaminoadenosine, and 2,4,6,8,10-dodecapentaenal, as dominant compounds in the methanolic extract of the same plant species [31].

The newly synthesized *D. flabellifolia*-AgNPs were visualized by SEM and analyzed by Image J software and are spherical with an average nanoparticles size about 16.961 ± 2.914 nm. After 70 min of boiling at 80°C. The formation of brown final color of the reaction mixture within 70 mn indicates the reduction of Ag+ ion ton Ag0 indicates the green synthesis of *D. flabellifolia*-AgNPs in the solution. It was previously reported that the appearance of brown color in the reaction mixture of AgNPs solution



Fig. 6. *In-vitro* inhibition of *C. violaceum* (A), *P. aeruginosa* (B), and *S. marcescens* (C) by *D. flabellifolia*-AgNPs.

was due to the excitation of surface plasmon vibrations [42]. Moreover, the formation of *D. flabellifolia*-AgNPs was confirmed by ultraviolet-visible spectroscopy technique, with a maximum surface plasmon resonance (SPR) absorption band between 430 nm-440 nm at 80°C. Few studies have reported the synthesis of silver nanoparticles using Ducrosia plant species extracts. In fact, using D. anethifolia extract, Amin and colleagues successfully synthesized spherical silver nanoparticles with a size range between 4 and 42.13 nm [43]. The average size of our D. flabellifolia-AgNPs was about 16.961±2.914 nm, closely related to the average size reported by Amin and colleagues [43] for D. anethifolia-AgNPs (11.4 nm). Similarly, using the same plant species (D. anethifolia), Darvish and colleagues [44] succeeded in synthesizing silver nitrate nanoparticles with a maximum absorbance at 460 nm and side distribution in the range of 3.02 to 20.8 nm (Average size 9.41 nm).

In this study, we report that green synthesized silver nanoparticles from *D. flabellifolia* aqueous extract were able to inhibit the growth of several pathogenic bacteria, mainly ESKAPE microorganisms with low MICs and MBCs values. In fact, our results that *D. flabellifolia* silver nanoparticles exhibited high to very high anti-ESKAPE activity with mean growth inhibition zone higher than 11

to 20 mm as defined by Parveen and colleagues [43]. In addition, our results showed that the MICs values ranged from 78 to 312 μ g/mL, and MBCs values from 312 to 625 µg/mL. While, against Candida species, D. flabellifolia-AgNPs were active at MIC value of about 321 µg/mL, and MFC value about 5 mg/mL. Similar results were reported by Amin and colleagues [44] who found that D. anethifolia silver nanoparticles were effective against S. aureus ATCC 29933, E. faecalis ATCC 51299, E. coli 25922, and P. aeruginosa ATCC 27853 at MICs values ranging from 32 to 128 µg/mL. Darvish and colleagues [45] reported that spherical silver nanoparticles synthesized using D. anethifolia aqueous extract were able to attenuate the viability of MCF-7 and MDA-MB-231 cell lines using the MTT technique in a time-dependent and concentration-dependent manner. In fact, the calculated inhibitory concentration of D. anethifolia-AgNPs (IC₅₀ 72h) after 72 h was estimated at about 16.87±2.76 µg/mL against MDA-MB-231 cell lines, and about $27.9\pm2.06 \,\mu\text{g/mL}$ about MCF-7 cell lines.

Previous results have reported the green synthesis of silver nanoparticles by using plant species and their waste material [33; 46-48]. It has been reported that silver nanoparticles synthesized by using extracts from plants, fungi, bacteria, actinomycetes, and algae (Red, blue, green, and brown) are active against large collection of Grampositive and Gram-negative bacteria including Bacillus subtilis, S. aureus, P. aeruginosa, B. cereus, Salmonella typhimurium, E. coli, Vibrio fluvialis, V. damsela, Proteus mirabilis, K. pneumoniae, methicillin-resistant S. aureus, and Micrococcus luteus [49]. More recently, Merghni and colleagues reported the green synthesis of silver nanoparticles by using dried orange peel extract with high antagonist activities against methicillin-resistant S. aureus strains with inhibition zones between 12 and 14 mm and MICs values between 1.56 and 12.5 µg/mL. The same authors reported that orange peel-AgNPs were able to disrupt the mature biofilm formed by MRSA strains on polystyrene microtiter plates. In addition, many researchers have reported the effectiveness of plant silver nanoparticles in combating ESKAPE pathogens [49-54]. In fact, Musthafa and colleagues [52] reported the new synthesis of spherical Ag-NPs using Picrorhiza kurroa plant extracts able to inhibit the growth of E. coli, K. pneumonia, P. aeruginosa, S. aureus, A. baumannii and E. aerogenes with growth inhibition zones ranging from 8 to 20 mm. More recently, Raza and colleagues reported that AgNPs synthesized using tea extracts have high antibacterial properties against ESKAPE pathogens causing approximately 80% bacterial cell death in within only three hours at a concentration of 0.1 mg/ mL as compared to ampicillin.

We report also in this study that the newly synthetized *D-flabellifolia* AgNPs were able to inhibit cell to cell communication in *C. violaceum*, *P. aeruginosa*, and *S. marcescens* at 125 µg/mL. Previous results have shown that silver nitrate nanoparticles synthetized by using *Vetiveria zizanioides* root extract were able to inhibit many virulence factors in *S. marcescens* controlled by the quorum sensing system like the production of prodigiosin, protease, lipase, and exopolysaccharide, and biofilm formation without inhibiting its growth [55]. Similarly, newly synthesized silver nanoparticles using piper betle aqueous extract were able to attenuate the production of QS-mediated virulence factors by uropathogenic bacteria like *Proteus mirabilis*, *P. aeruginosa*, *E. coli*, and *S. marcescens*

[56]. Recently, Qais and colleagues [57] reported that silver nanoparticles synthesized using aqueous extract from *Carum copticum* were able to inhibit violacein production in *C. violaceum*, the production of pyoverdine, proteases, elastases, rhamnolipid, and motility in *P. aeruginosa*, and prodigiosin, proteases, and motility in *S. marcescens. In silico* approaches highlighted that AgNPs could bind to LasI synthase, RhII synthase, and transcriptional receptor protein LasR and RhIR of the two *P. aeruginosa* Las and RhI quorum sensing systems [58]. Similar results are reported with selenium nanoparticles [59], and silver, zinc oxide, and copper oxide nanoparticles [60].

5. Conclusion

Overall, the newly synthesized nanoparticles using aqueous extract from *D. flabellifolia* aerial parts showed a spherical shape with low particle size. The obtained extract was dominated by phytoconstituents with promising biological activities. Results showed that *D. flabellifolia* silver nanoparticles were able to interfere with the quorum sensing systems in *P. aeruginosa*, *C. violaceum*, and *S. marcescens* bacteria. Similarly, the newly synthesized silver nanoparticles possessed potent anti-ESKAPE activities. The results highlight the possible use of silver nitrate nanoparticles synthesized using *D. flabellifolia* extract as an alternative solution to combat multidrug-resistant microorganisms and to attenuate their virulence.

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Conflict of interest

The authors declare no conflict of interest.

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